Intramolecular Cyclizations of α-Stannyl Radicals to Acylsilanes: Regiospecific Syntheses of Five-membered Cyclic Silyl Enol Ethers
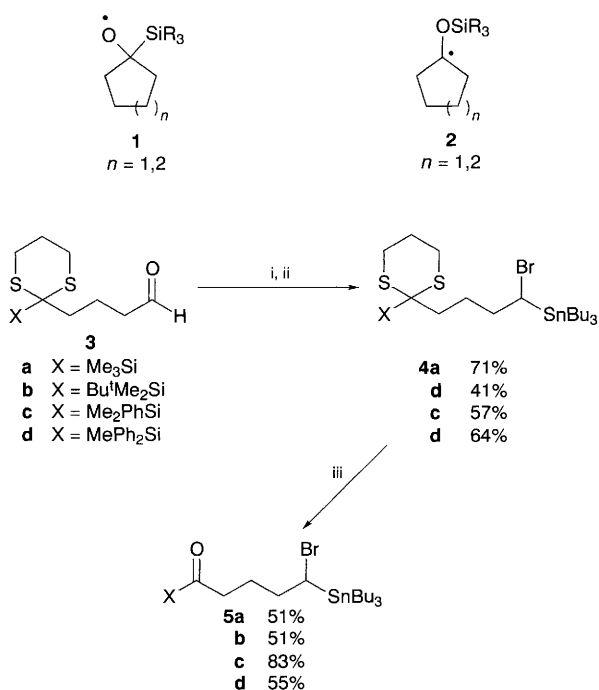
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α-Stannyl radicals generated from acylsilanes **5**, **13** and **14** cyclize to give good yields of cyclic silyl enol ethers after sequential cyclizations, radical Brook rearrangements and β-scissions.

Acylsilanes have gained more and more attention in recent years.[1] Radical cyclization reactions of acylsilanes involve the formation of β-silyl alkoxy radicals **1** which quickly rearrange to give α-silyloxy radicals **2**.[2] This special feature creates opportunities for further elaboration.[3] Here we report the use of α-stannyl radicals to cyclize with acylsilanes and give cyclic silyl enol ethers in a regiospecific way.

Aldehydes **3**[3a] were treated with tributyltin anions and the resulting α-stannyl alcohols were converted to α-stannyl bromides **4** using triphenylphosphine and tetrabromomethane (Scheme 1). Hydrolysis of **4** with ceric ammonium nitrate (CAN)[4] gave acylsilanes **5** in moderate yields over the three steps.

Cyclizations of **5** in refluxing benzene with slow addition of a catalytic amount of tributyltin hydride (0.15 equiv.) and AIBN (0.05 equiv.) gave silyl enol ethers **6** (Scheme 2).† The presence of **6** was confirmed by comparing their crude ¹H NMR spectra with those of the authentic samples prepared from cyclopentanone. However, the crude products of **6** were converted to the 2,4-dinitrophenylhydrazone of cyclopentanone to estimate the yields of the cyclizations. In the cases of **6a**, **c** and **d**, the overall conversion yields were very good. The low yield in the case with a bulky *tert*-butyldimethylsilyl group **6b** indicated that the cyclization was not efficient. This may attributable to the steric interactions between the silyl group and the tributylstannyl group. The whole process invovles the generation of α-stannyl radicals **8** which cyclize to give radicals **9**. These radicals undergo radical Brook rearrangements[5] to give radicals **10** which are set up to perform facile β-scissions and give silyl enol ethers **6** with concomitant formation of a tributyltin radical. The tin radical then abstracts the bromine atom from **5** to regenerate radicals **8**, thus completes the cycle.
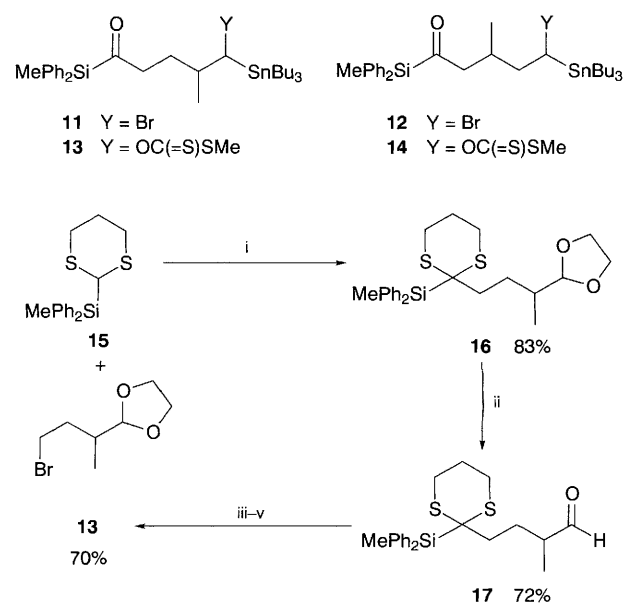
To demonstrate the regiospecific nature of this approach for the preparation of cyclic silyl enol ethers, we intended to synthesize bromides **11** and **12**. However, the same route used for the synthesis of **5** gave low yields in these cases. This was probably due to the instability of the structures of type **4** in which a nucleophile and an electrophile coexist.[6] Therefore, a different route was developed. Alkylation of the 2-silyl-1,3-dithiane **15** with 2-(3-bromo-1-methylpropyl)-1,3-dioxolane[7] gave **16** in 83% yield (Scheme 3). The aldehyde **17** obtained from hydrolysis of **16** was treated with tributyltin lithium, and the resulting alkoxide was trapped sequentially with carbon disulfide and methyl iodide to construct the xanthate moiety. The dithiane was then hydrolysed with
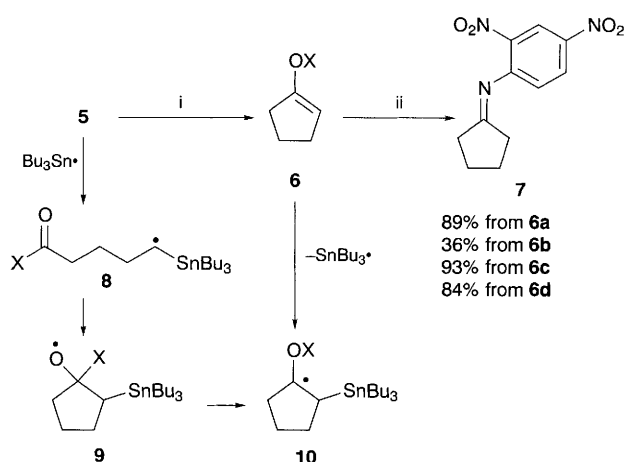
Scheme 2 *Reagents and Conditions*: i, Bu₃SnH (0.15 equiv.), AIBN (0.05 equiv.), C₆H₆, 80 °C; ii, 2,4-dinitrophenylhydrazine, H⁺, EtOH

89% from **6a**
36% from **6b**
93% from **6c**
84% from **6d**

11 Y = Br
13 Y = OC(=S)SMe

12 Y = Br
14 Y = OC(=S)SMe

16 83%

17 72%

13
70%

Scheme 3 *Reagents*: i, BuLi; ii, *p*-Me-C₆H₄SO₃H (cat.), H₂O, THF; iii, Bu₃SnLi; iv, CS₂, MeI; v, (CF₃CO₂)₂IC₆H₅, H₂O, MeCN, THF, NaHCO₃

1
n = 1,2

2
n = 1,2

3

a   X = Me₃Si
b   X = BuᵗMe₂Si
c   X = Me₂PhSi
d   X = MePh₂Si

4a   71%
d   41%
c   57%
d   64%

5a   51%
b   51%
c   83%
d   55%

Scheme 1 *Reagents*: i, Bu₃SnLi; ii, Ph₃P, CBr₄; iii, CAN, MeCN, H₂O

iodobenzene bis(trifluoroacetate) to give **13** in 70% yield from the aldehyde.[8] The xanthate **14** was synthesized accordingly in similar yields.

Radical cyclizations of **13** and **14** under the same conditions as stated above gave the two regioisomeric silyl ethers **18** and **19**, respectively. To confirm the structures, the crude products were treated with phenylselenyl bromide in dichloromethane at −78 °C. The selenide **20** was obtained in 68% yield from **13** as a mixture of *cis/trans* (1:2) isomers.[9] The selenide **21** was obtained in 77% yield from **14** as an 8.8:1 mixture of two stereoisomers.‡

Attempted radical cyclization of **22** initiated with tributyltin hydride (0.15 equiv.) and AIBN (0.05 equiv.) was not successful. Only the reduction product **23** was observed along with mostly unreacted **22**. With a full equivalent of tributyltin hydride, the acylsilane **23** was isolated in 80% yield. It has been reported for radical cyclizations of acylsilanes that 1,5-hydrogen atom transfer was a competing process with 1,6-cyclization.[3a] In the case of **22**, the unfavourable interactions between the silyl and stannyl groups probably blocked the cyclization completely.

In summary, the radical cyclizations described herein represent a novel regiospecific approach to the syntheses of five-membered cyclic silyl enol ethers. This report also demonstrates that α-stannyl radicals can be very useful if properly designed. The potential of α-stannyl radicals is currently under investigations.

Footnotes

† *Typical procedure*: Tributyltin hydride (21 µl, 0.078 mmol) and AIBN (5 mg, 0.03 mmol) in benzene (5 ml) were added over 1 h to a solution of **5c** (295 mg, 0.503 mmol) in benzene (5 ml) heated at 80 °C. The resulting solution was heated for a further hour and then concentrated *in vacuo* to obtain the crude product which was used directly in the subsequent reaction.

‡ The major isomer was presumably the *trans* isomer; however, the stereochemistry was not determined. The characteristic ^{1}H NMR (400 MHz, CDCl$_3$) signal for the major isomer appears at δ 3.82 (br d, *J* 6.6 Hz), and that of the minor one appears at 3.72 (br dd, *J* 11, 8.4 Hz).

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18   **19**   **20**   **21**

22   **23**